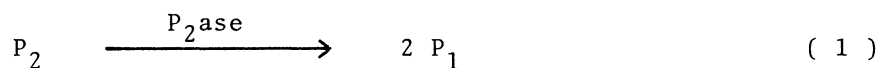


MEASUREMENT OF ENZYMIC ACTIVITY OF INORGANIC PYROPHOSPHATASE  
FOR PYROPHOSPHATE BY FLOW INJECTION ANALYSISNorimasa YOZA,\* Hisanobu HIRANO, Mayumi OKAMURA, Shigeru OHASHI,  
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A flow injection system was designed for the spectrophotometric measurement of enzymatic hydrolysis of pyrophosphate by inorganic pyrophosphatase ( EC 3.6.1.1 ). Rapid hydrolysis with a very short half-life ( .2 min ) could be easily monitored by the selective detection of orthophosphate using a Mo(VI) reagent.

Inorganic pyrophosphatase ( EC 3.6.1.1, P<sub>2</sub>ase ) is an enzyme that catalyzes the hydrolytic conversion of inorganic pyrophosphate ( diphosphate, P<sub>2</sub> ) to orthophosphate.



Extensive investigations<sup>1-6)</sup> by biochemists have been reported on the catalytic properties of P<sub>2</sub>ase for P<sub>2</sub>, because the enzyme P<sub>2</sub>ase is known to be widely distributed in nature and may play an important role in cell metabolism.

Our major concerns are the catalytic activity of P<sub>2</sub>ase for phosphorus compounds in environmental water and the utility of P<sub>2</sub>ase in inorganic phosphorus chemistry. One of important tasks in these fields is the development of automated techniques to be used for the rapid measurement of enzymatic activities. This paper describes an analytical method based on flow injection analysis ( FIA )<sup>7)</sup> by which the rate of enzymatic hydrolysis of P<sub>2</sub> with a half-life less than a few minutes can be easily measured.

The FIA manifold in Fig. 1 was designed so that the enzymatic reaction in

Eq. 1 could be stopped instantaneously and the product  $P_1$  could be detected selectively in the presence of the substrate  $P_2$ . Each flow rate of water and Mo(VI) reagent (  $0.03 \text{ M Mo(VI)} + 0.3 \text{ M H}_2\text{SO}_4 + 0.01 \text{ mM } P_1$ ;  $1 \text{ M} = 1 \text{ mol dm}^{-3}$  ) was adjusted to  $1.0 \text{ ml/min}$  by using a reciprocating pump ( Kyowa KHU-W-52 or Jasco RP-4 ). A sample solution ( S ) was injected into the water stream via a loop-valve injector ( Seishin VMU-6,  $100 \mu\text{l}$  loop ). Only  $P_1$  reacted with the Mo(VI) reagent in the reaction coil ( RC,  $0.5 \text{ mm ID} \times 5 \text{ m PTFE}$ ,  $30 \text{ }^\circ\text{C}$  ) to form molybdophosphoric acid. The absorbance of this yellow complex was monitored with a flow-through cell attached to a spectrophotometer ( D, Hitachi 200-10 or Jasco UVIDEK-320 ).<sup>8)</sup> A back-pressure coil ( BC,  $0.25 \text{ mm ID} \times 2 \text{ m}$  ) was located at the exit of the cell. The addition of  $P_1$  to the Mo(VI) reagent is recommended to stabilize the base-line level if molybdophosphoric acid tends to be adsorbed on the flow system.

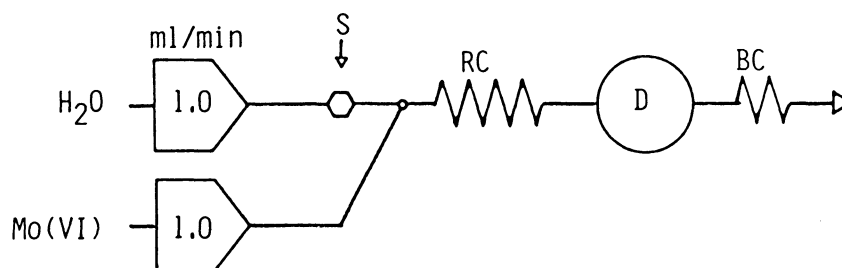


Fig. 1. FIA manifold for the selective determination of  $P_1$ .

Standard samples of  $P_1$  (  $\text{KH}_2\text{PO}_4$  ),  $P_2$  (  $\text{Na}_4\text{P}_2\text{O}_7 \cdot 10\text{H}_2\text{O}$  ), and  $P_3$  ( triphosphate,  $\text{Na}_5\text{P}_3\text{O}_{10}$  ) were injected into the manifold ( Fig. 1 ) at 1 min intervals. The residence time in the reaction coil was about 40 s. Calibration profiles are shown in Fig. 2. Orthophosphate could be determined reproducibly with a relative standard deviation of less than 1% in the dynamic range of  $0.01 \text{ mM} - 1 \text{ mM } P_1$ . This working range was satisfactory for the activity measurement of  $P_2$ ase in most cases. On the other hand polymeric  $P_2$  and  $P_3$  did not show positive signals. The negative signals observed at  $10 \text{ mM}$  may be ascribed to a matrix effect.

$P_2$ ase was allowed to react with  $P_2$  at  $30 \text{ }^\circ\text{C}$  in a buffered medium ( pH 7.2,  $5 \text{ mM Tris-HCl}$  buffer ) containing  $1 \text{ mM MgCl}_2$  as an activator. Aliquots of the reaction mixture ( each  $100 \mu\text{l}$  ) were successively injected at 1 min intervals into the manifold ( Fig. 1 ) to monitor the reaction product  $P_1$ . Figure 3 shows kinetic FIA profiles obtained by keeping the substrate concentration constant,  $[P_2] = 0.5 \text{ mM}$ , and varying the enzyme concentrations. The concentrations of  $P_2$ ase from yeast ( Sigma ) were estimated on the basis of the protein content specified and the

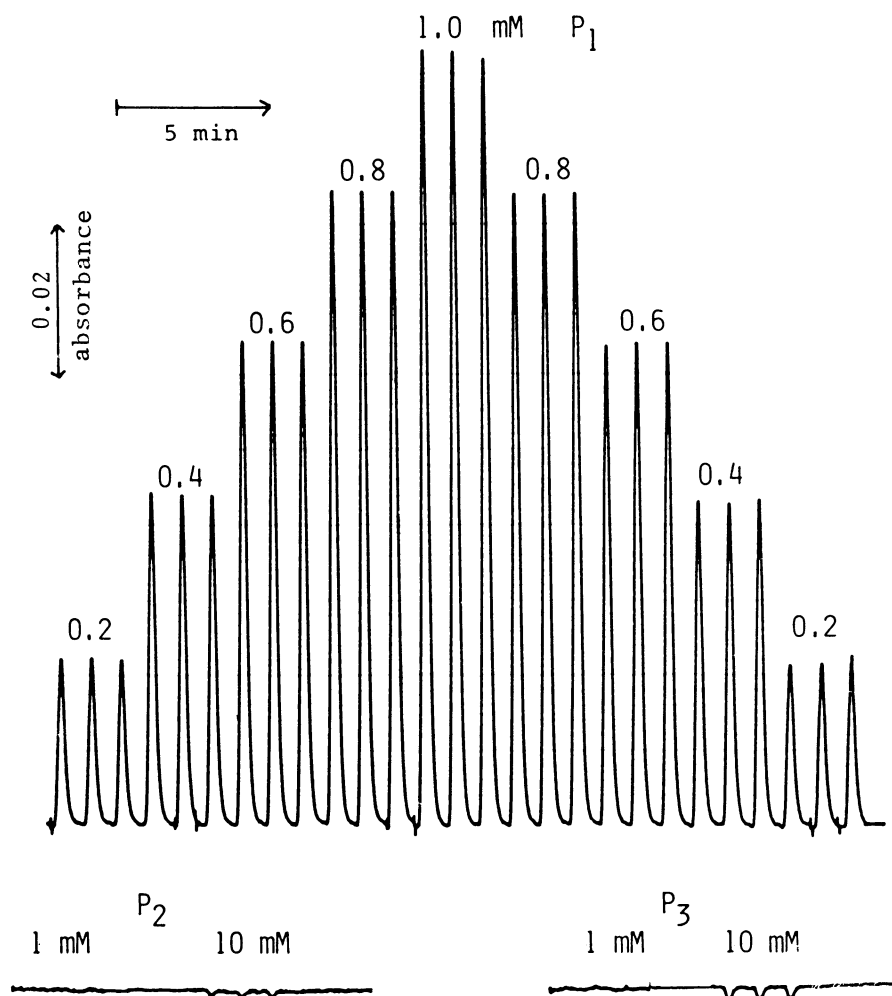


Fig. 2. FIA calibration profiles for P<sub>1</sub>, P<sub>2</sub>, and P<sub>3</sub>.

molecular weight ( 64000 ) in the literature.<sup>4)</sup> The peak height or the amount of P<sub>1</sub> in each FIA profile increased linearly with time and then tended to become constant toward the end of the enzymatic reaction. The slopes in FIA profiles or the reaction rates were proportional to the enzyme concentrations, but independent of the substrate concentrations ( 0.2 mM, 0.5 mM, and 1 mM P<sub>2</sub> ). The observed reaction rate,  $V$ , is considered to be the maximum reaction rate,  $V_{\max}$ , in the well-known Michaelis-Menten expression<sup>9)</sup>;  $V = V_{\max} [P_2]_0 / ( [P_2]_0 + K_m )$ . The Michaelis constant  $K_m$  for P<sub>2</sub>ase has been shown to be as low as 0.008 mM.<sup>2)</sup>

It was found that the enzymatic hydrolysis of P<sub>2</sub> of a very short half-life ( 2.2 min ) can be measured by FIA as shown in Fig. 3 ( c ). The specific activity of P<sub>2</sub>ase was estimated from the kinetic FIA profiles to be 730 U./mg, i.e., 1 mg protein liberated 730  $\mu$ mol of P<sub>1</sub> per minute. This value is somewhat higher than

500-600 U/mg specified by Sigma, but is in agreement with 700 U/mg given by Mel'nik et al.<sup>2)</sup>

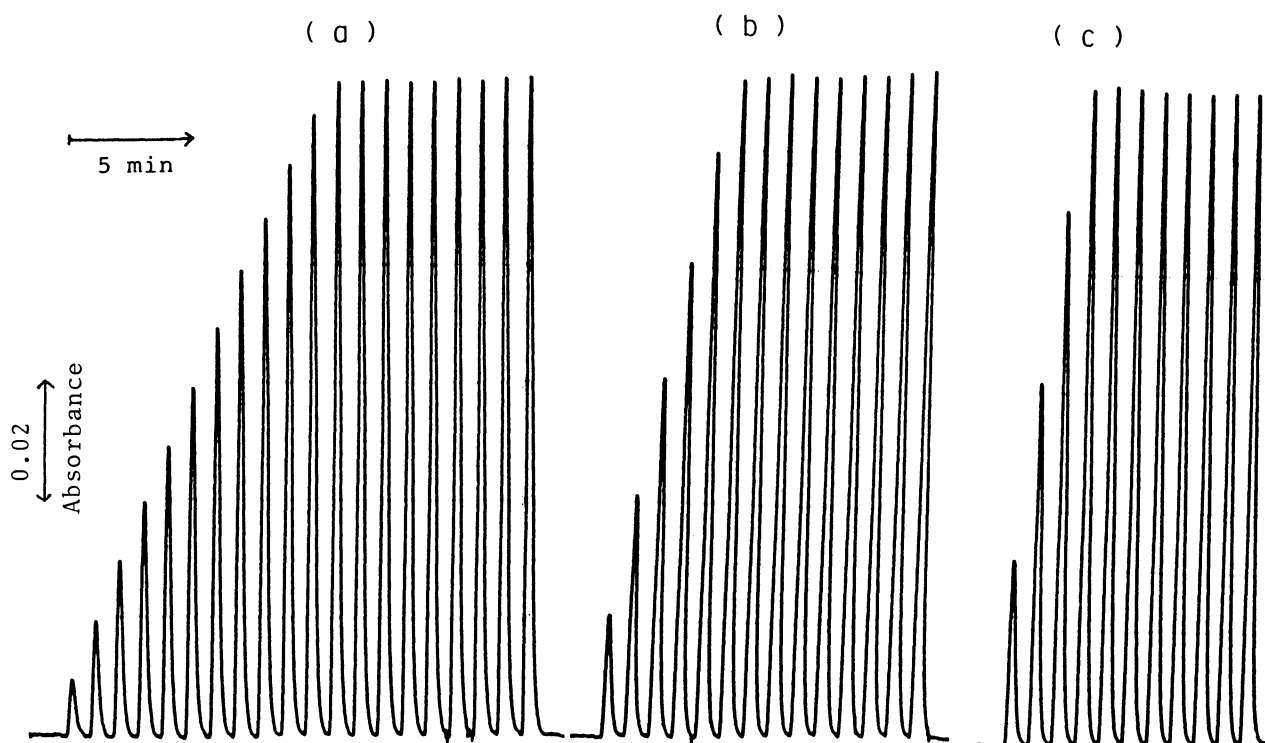


Fig. 3. Kinetic FIA profiles for enzymatic hydrolysis of pyrophosphate.  
 $P_2ase$  concentrations; ( a )  $1.5 \times 10^{-9}$  M, ( b )  $3.0 \times 10^{-9}$  M  
 and ( c )  $4.5 \times 10^{-9}$  M.

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